

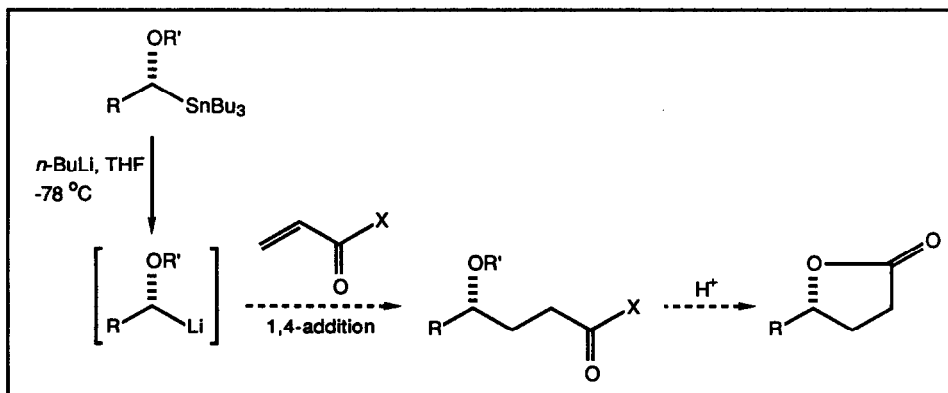
AN EXPEDITIOUS ENANTIOSELECTIVE SYNTHESIS OF γ -LACTONES

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Summary: α -Alkoxyorganolithium reagents react with α,β -unsaturated trimethylhydrazides to give γ -alkoxytrimethylhydrazides in moderate to good yields. Acid hydrolysis of the latter compounds yields γ -lactones. A short enantioselective route to (*S*)-hexanolide is described.

The lactone functionality is prevalent among many natural products.¹ For example, certain γ -lactones are pheromones for several insect species. In addition, the food industry utilizes some lactones as flavouring components. We envisioned an expeditious synthesis of γ -lactones² by the 1,4-addition of α -alkoxyorganolithium reagents³ to α,β -unsaturated carboxylic acid derivatives followed by cyclization of the resulting γ -alkoxyacid derivatives. α -Alkoxyorganolithium reagents can be obtained by transmetalation of α -alkoxyorganostannanes.^{3a} Moreover, α -alkoxyorganostannanes can be prepared in enantiomerically enriched form.⁴ Thus, an enantioselective route to γ -lactones is feasible as well.

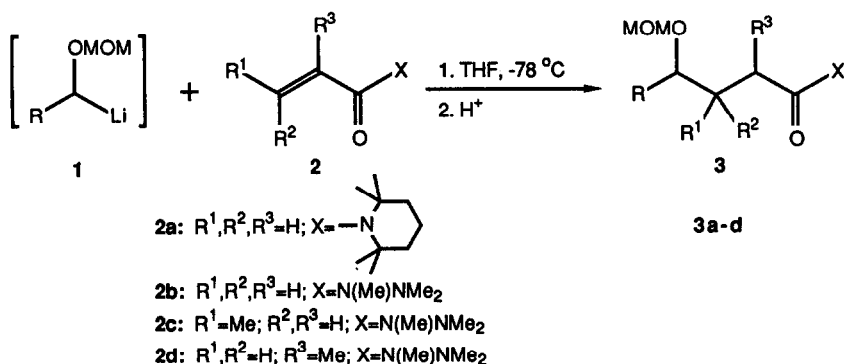


Initial investigations centred on the addition of racemic α -alkoxyorganolithium compounds⁵ to α,β -unsaturated amides. Good yields of 1,4-addition product were obtained with the highly hindered acrylic acid tetramethylpiperidide (2a)⁶ as the Michael acceptor (Table).⁷ Unfortunately, attempts to cyclize γ -alkoxyamide 3a ($\text{R} = n\text{-Bu}$) to the γ -lactone under acidic conditions (e.g. 3 N HCl, *p*-TsOH, HClO_4) were uniformly unsuccessful with no lactone observed.

It is known that alkylolithiums add to α,β -unsaturated trimethylhydrazides in a conjugate fashion.⁸ Thus, we examined the reaction of α -alkoxyorganolithiums with various α,β -unsaturated trimethylhydrazides⁹ (Table).

Modest yields of 1,4-addition products were obtained with acrylic acid trimethylhydrazide (2b) with the principal side reaction being the oligomerization of the intermediate hydrazide lithium enolate with a molecule of 2b. Yields of 1,4-addition product obtained with crotonic acid trimethylhydrazide (2c) were poor to moderate as competing 1,2-addition was occurring.¹⁰ However, much better yields of 1,4-addition product were obtained with methacrylic acid trimethylhydrazide (2d); presumably, oligomerization of the intermediate lithium enolate is reduced due to the relatively hindered α -carbon. Not surprisingly, only modest diastereoselection was observed in additions of α -alkoxyorganolithiums to 2c and 2d.

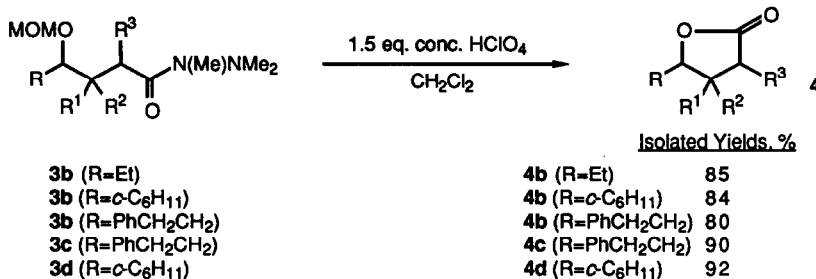
Table. 1,4-Addition of α -Alkoxyorganolithium Reagents to 2.



R	Isolated Yields, % ^a			
	3a	3b	3c ^{b,c}	3d ^b
Et	75	50	49 (51:49)	70 (67:33)
<i>t</i> -Pr	65	49	37 (54:46)	65 (71:29)
<i>n</i> -Bu	68	40	53 (71:29)	71 (61:39)
α -C ₆ H ₁₁	65	42	37 (72:28)	73 (66:34) ^d
PhCH ₂ CH ₂	69	52	61 (60:40) ^d	70 (61:39)

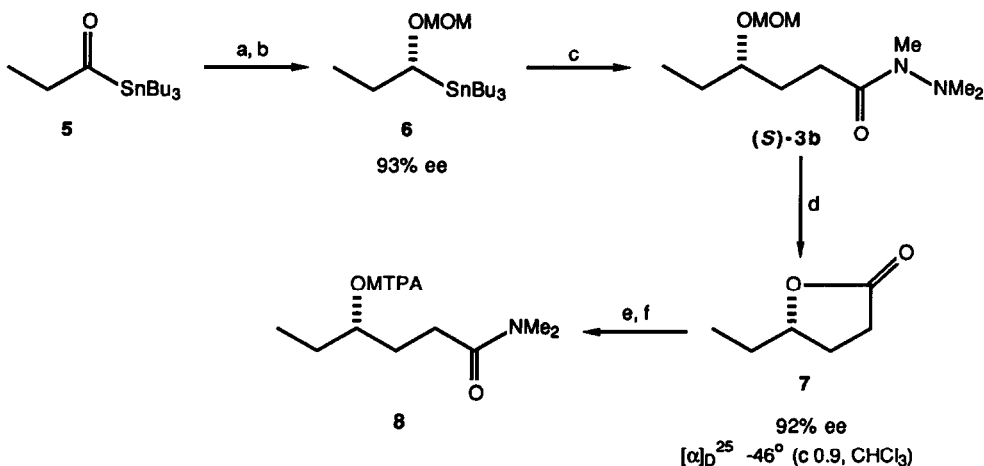
^a All new compounds exhibited satisfactory spectroscopic and analytical data. ^b Numbers in parentheses refer to the diastereomeric ratio determined by GC analyses. ^c 16-40% yield of 1,2-addition product observed. ^d The major diastereomer was assigned as *syn* on the basis of its conversion to the γ -lactone. The chemical shift of H_5 of the γ -lactone was diagnostic.² For 4c ($R = PhCH_2CH_2$), *cis* $\delta_{H_5} = 4.41$ ppm, *trans* $\delta_{H_5} = 3.99$ ppm. For 4d ($R = \alpha$ -C₆H₁₁), *cis* $\delta_{H_5} = 4.23$ ppm, *trans* $\delta_{H_5} = 4.05$ ppm.

In contrast to the tetramethylpiperidides **3a**, cyclizations of the resulting γ -alkoxyhydrazides **3b-3d** to the γ -lactones **4** proceeded smoothly and in good yields. Exposure of the hydrazide to HClO_4 (1.5 eq) in CH_2Cl_2 (rt, 8 h) afforded the expected γ -lactone as the exclusive product. Some results are tabulated below. Hence, although yields of 1,4-addition product are only moderate, the overall synthesis of substituted γ -lactones is expedient (two steps) and simple.



Attention was then turned to the enantioselective synthesis of a γ -lactone (Scheme). Retention of configuration in the 1,4-addition of an α -alkoxyorganolithium reagent to an α,β -unsaturated acid derivative has not been previously demonstrated. Starting with the enantiomerically enriched α -alkoxyorganostannane **6**,¹¹ (*S*)-hexanolide¹² (**7**) was prepared in two steps. The enantiomeric purity of **7** was determined by its optical rotation¹³ and by ¹³C NMR analysis of Mosher ester **8**,¹⁴ obtained by derivatization of **7**. Since the enantiomeric purities of **6** and **7** are the same, then the conjugate addition of the α -alkoxyorganolithium reagent to the unsaturated trimethylhydrazide occurred with retention of configuration.

Scheme. Enantioselective Synthesis of (*S*)-Hexanolide (**7**).



Key: (a) (*S*)-BINAL-H, THF, -78 °C. (b) MOM-Cl, *i*-Pr₂NEt, CH₂Cl₂, 59% from **5**.
 (c) *n*-BuLi, THF, -78 °C, 10 min then **2**, 1 h, 50%.
 (d) 1.5 eq conc. HClO₄, CH₂Cl₂, rt, overnight, 85%.
 (e) Me₂NH, Et₃N, -20 °C → RT. (f) (*R*)-MTPA-Cl, Et₃N, DMAP, CH₂Cl₂.

Thus, we have described an efficacious route to substituted γ -lactones and to enantiomerically enriched γ -alkyl- γ -lactones. The latter compounds may be prepared in four steps (ca. 25% overall yield) from acylstannanes. Of the previously described enantioselective routes to γ -alkyl- γ -lactones,¹⁵ that of Silverstein's is perhaps the most general and economical (four steps from *D*- or *L*-glutamic acid, ca. 9-19% overall yields).¹ However, it is incapable of accessing branched γ -alkyl- γ -lactones (e.g. 3b, R=*i*-Pr or *c*-C₆H₁₁) since the key step involves displacement of a primary tosylate (i.e. 4, R=CH₂OTs; R¹,R²,R³=H) with an organocuprate reagent. The present method allows one to prepare straight-chain and branched γ -alkyl- γ -lactones enantioselectively.

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References and Notes

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10. In fact, with hydrazide 2e [R¹=TMS; R²,R³=H; X=N(Me)NMe₂], only the product of 1,2-addition was observed with 1 (R=*i*-Pr) (49% yield). With *n*-BuLi, 2e affords only the 1,4-addition product (66% yield). Perhaps stabilization of the tetrahedral intermediate arising from 1,2-addition by the alkoxy group [cf: additions to *N*-methoxyamides (Weinreb, S. M.; Nahm, S. *Tetrahedron Lett.* **1981**, *22*, 3815)] causes 1,2-addition to be more favourable when α -alkoxyorganolithiums 1 are used.
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